



# Voltage and Current

# 2.2 – Atoms and Their Structure

♃ Nucleus

♃ Protons

♃ Electrons

♃ Neutrons

# Atoms and Their Structure

♋ Shells and subshells of the atomic structure

♋ Free electrons

## 2.3 - Voltage

- ⚡ The flow of charge is established by an external “pressure” derived from the energy that a mass has by virtue of its position: **Potential energy**
- ⚡ Energy: the capacity to do work
  - ⚡ If a mass ( $m$ ) is raised to some height ( $h$ ) above a reference plane, it has a measure of potential energy expressed in joules (J) that is determined by
  - ⚡  **$W$  (potential energy) =  $mgh$**   
where  $g$  is the gravitational acceleration ( $9.8 \text{ m/s}^2$ )

# Voltage

- ⚡ A potential difference of 1 volt (V) exists between two points if 1 joule (J) of energy is exchanged in moving 1 coulomb (C) of charge between the two points
- ⚡ The unit of measurement **volt** was chosen to honor Alessandro Volta

# Voltage

- ⚡ A potential difference or voltage is always measured between two points in the system. Changing either point may change the potential difference between the two points under investigation.
- ⚡ Potential difference between two points is determined by:  $V = W/Q$  (volts)

# Voltage

- ⌘ Notations for sources of voltage and loss of potential
  - ⌘ E - Voltage sources (volts)
  - ⌘ V - Voltage drops (volts)
- ⌘ Potential – The voltage at a point with respect to another point in the electrical system. Typically the reference point is the ground, which is at zero potential.

# Voltage

- ⚡ Potential difference: The algebraic difference in potential (or voltage) between two points of a network.
- ⚡ Voltage: When isolated, like potential, the voltage at a point with respect to some reference such as ground.
- ⚡ Voltage difference: The algebraic difference in voltage (or potential) between two points of a system. A voltage drop or rise is as the terminology would suggest.
- ⚡ Electromotive force (emf): The force that establishes the flow of charge (or current) in a system due to the application of a difference in potential.



# Voltage

## Summary

- ⚡ The applied potential difference (in volts) of a voltage source in an electric circuit is the “pressure” to set the system in motion and “cause” the flow of charge or current through the electrical system.

## 2.4 - Current

- ⌘ The free electron is the charge carrier in a copper wire or any other solid conductor of electricity
- ⌘ With no external forces applied, the net flow of charge in a conductor in any one direction is zero
- ⌘ Basic electric circuit

# Current

- ⚡ Safety considerations
  - ⚡ Even small levels of current through the human body can cause serious, dangerous side effects
  - ⚡ Any current over 10 mA is considered dangerous
  - ⚡ currents of 50 mA can cause severe shock
  - ⚡ currents over 100 mA can be fatal
  - ⚡ Treat electricity with respect – not fear

# 2.5 – Voltage Sources

- ♋ dc – Direct current
  - ♋ Unidirectional (“one direction”) flow of charge
  - ♋ Supplies that provide a fixed voltage or current

# Voltage Sources

- ⌘ dc Voltage sources
  - ⌘ Batteries (chemical action)
  - ⌘ Generators (electromechanical)
  - ⌘ Power supplies (rectification)

# Voltage Sources

- ⚡ Batteries: combination of two or more similar cells
  - ⚡ A cell being a fundamental source of electrical energy developed through the conversion of chemical or solar energy
  - ⚡ All cells are divided into Primary and Secondary types
    - ⚡ Primary type is not rechargeable
    - ⚡ Secondary is rechargeable; the cell can be reversed to restore its capacity
    - ⚡ Two most common rechargeable batteries are the lead-acid unit (primarily automotive) and the nickel-cadmium (calculators, tools, photoflash units and shavers)

# Voltage Sources

- ⌘ Each cell establishes a potential difference at the expense of chemical energy and each has the following components:
  - ⌘ Positive electrode
  - ⌘ Negative electrode
  - ⌘ Electrolyte (the contact element and the source of ions for conduction between terminals)

# Voltage Sources

- ⌘ Alkaline primary cells
  - ⌘ Powered zinc anode (+)
  - ⌘ Potassium (alkali metal) electrolyte
  - ⌘ Manganese dioxide, carbon cathode (–)



# Voltage Sources

- ⚡ Lead-acid secondary cell
  - ⚡ Sulfuric acid is the electrolyte
  - ⚡ The electrodes are spongy lead (Pb) and lead peroxide (PbO<sub>2</sub>)

# Voltage Sources

- ⚡ Nickel-cadmium secondary cell
  - ⚡ Rechargeable battery (Capable of 1,000 charge/discharge cycles)
  - ⚡ charged by a constant current source
- ⚡ Nickel-hydrogen and nickel-metal hydride secondary cells
  - ⚡ Nickel-hydrogen cell currently limited primarily to space vehicles
  - ⚡ Nickel-metal hydride cell is actually a hybrid of the nickel-cadmium and nickel-hydrogen cell – Expensive, but it is a valid option for applications such as portable computers

# Voltage Sources

## ♋ Solar cell

- ♋ A fixed illumination of the solar cell will provide a fairly steady dc voltage for driving loads from watches to automobiles
- ♋ Conversion efficiencies are currently between 10% and 14%

# Voltage Sources

## ⚡ Ampere-hour rating

- ⚡ Batteries have a capacity rating in ampere-hours
- ⚡ A battery with an ampere-hour rating of 100 will theoretically provide a steady current of 1A for 100 h, 2A for 50 h or 10A for 10 h
- ⚡ Factors affecting the rating: rate of discharge and temperature
  - ⚡ The capacity of a dc battery decreases with an increase in the current demand
  - ⚡ The capacity of a dc battery decreases at relatively (compared to room temperature) low and high temperatures

# Voltage Sources

## ⌘ Generators

⌘ Voltage and power-handling capabilities of the dc generator are typically higher than those of most batteries, and its lifetime is determined only by its construction

## ⌘ Power supplies

⌘ The dc supply encountered most frequently in the laboratory employs the rectification and filtering processes as its means toward obtaining a steady dc voltage

# Voltage Sources

## ⌘ dc Current sources

- ⌘ The current source will supply, ideally, a fixed current to an electrical/electronic system, even though there may be variations in the terminal voltage as determined by the system

# 2.8 Conductors and Insulators

⌘ Conductors are those materials that permit a generous flow of electrons with very little external force (voltage) applied

**In addition,**

⌘ Good conductors typically have only one electron in the valance (most distant from the nucleus) ring.

# Conductors and Insulators

- ⚡ Insulators are those materials that have very few free electrons and require a large applied potential (voltage) to establish a measurable current level
- ⚡ Insulators are commonly used as covering for current-carrying wire, which, if uninsulated, could cause dangerous side effects
- ⚡ Rubber gloves and rubber mats are used to help insulated workers when working on power lines
- ⚡ Even the best insulator will break down if a sufficiently large potential is applied across it



# Conductors and Insulators

✂ Table 2.1 shows the relative conductivity of various materials

**TABLE 2.1**  
*Relative conductivity of various materials.*

Metal	Relative Conductivity (%)
Silver	105
Copper	100
Gold	70.5
Aluminum	61
Tungsten	31.2
Nickel	22.1
Iron	14
Constantan	3.52
Nichrome	1.73
Calorite	1.44

✂ Table 2.2 shows breakdown strength of some common insulators

**TABLE 2.2**  
*Breakdown strength of some common insulators.*

Material	Average Breakdown Strength (kV/cm)
Air	30
Porcelain	70
Oils	140
Bakelite	150
Rubber	270
Paper (paraffin-coated)	500
Teflon	600
Glass	900
Mica	2000

## 2.9 Semiconductors

- ⚡ Semiconductors are a specific group of elements that exhibit characteristics between those of insulators and conductors
- ⚡ Semiconductor materials typically have four electrons in the outermost valence ring
- ⚡ Semiconductors are further characterized as being photoconductive and having a negative temperature coefficient
  - ⚡ Photoconductivity: Photons from incident light can increase the carrier density in the material and thereby the charge flow level
  - ⚡ Negative temperature coefficient: Resistance will decrease with an increase in temperature (opposite to that of most conductors)

# 2.10 Ammeters and Voltmeters

- ⌘ Ammeter (Milliammeter or Microammeter)
  - ⌘ Used to measure current levels
  - ⌘ Must be placed in the network such that the charge will flow through the meter
- ⌘ Voltmeter
  - ⌘ Used to measure the potential difference between two points

# Ammeters and Voltmeters

- ⌘ Volt-ohm-milliammeter (VOM) and digital multimeter (DMM)
  - ⌘ Both instruments will measure voltage and current and a third quantity, resistance
  - ⌘ The VOM uses an analog scale, which requires interpreting the position of the pointer on a continuous scale
  - ⌘ The DMM provides a display of numbers with decimal point accuracy determined by the chosen scale.

# 2.11 Applications

## ⌘ Flashlight

- ⌘ Simplest of electrical circuits

- ⌘ Batteries are connected in series to provide a higher voltage (sum of the battery voltages)

# Applications

## ♋ 12-V Car battery charger

- ♋ Used to convert 120-V ac outlet power to dc charging power for a 12-V automotive battery, using a transformer to step down the voltage, diodes to rectify the ac (convert it to dc), and in some cases a regulator to provide a dc voltage that varies with level of charge.

# Applications

- ⌘ Answering machines/Phones dc supply
  - ⌘ A wide variety of devices receive their dc operating voltage from an ac/dc conversion system
  - ⌘ The conversion system uses a transformer to step the voltage down to the appropriate level, then diodes “rectify” the ac to dc, and capacitors provide filtering to smooth out the dc.